

Nagoya University, Furo-cho, Chikusa-ku, Nagoya, Japan. Isothermal electrical conductivity measurements on niobium oxides were carried out over the temperature range from 1010 to 1300°C as a function of oxygen partial pressure in order to clarify the phase relations. Existence regions of the intermediate oxide phases between NbO<sub>2</sub> and Nb<sub>2</sub>O<sub>5</sub> were found from the discontinuities in electrical conductivity curves. These oxide phases were also analyzed by a gravimetric method and by X-ray diffractometry. From these results the phase diagram for this system is proposed. The defect structures of these phases are also discussed.

*Magnetochemical Properties of Tetranuclear Rhodoso and Pfeiffer Chromium(III) Complexes in a Series of Compounds.* HANS U. GÜDEL AND URS HAUSER, Institut für anorganische und physikalische Chemie, Universität Bern, CH-3000 Bern 9, Switzerland. Seven salts of the structurally related tetranuclear rhodoso and Pfeiffer chromium(III) complexes were prepared. Magnetic susceptibilities were measured and the energy splittings of the electronic ground state caused by exchange interactions were determined. There are marked differences in the low-temperature magnetic properties. Crystal packing and hydrogen bonding effects are the most likely causes for the differences in the exchange-splitting pattern.

*Polymorphic Transformations of Bi<sub>2</sub>MoO<sub>6</sub>.* A. WATANABE AND H. KODAMA, National Institute for Researches in Inorganic Materials, 1-1 Namiki, Sakura-mura, Niihari-gun, Ibaraki, 305 Japan. The polymorphism of Bi<sub>2</sub>MoO<sub>6</sub> has been studied by differential thermal analysis, differential dilatometry, and differential scanning calorimetry with  $\gamma$ -form specimens having the kochlinite structure prepared by sintering the oxides Bi<sub>2</sub>O<sub>3</sub> and MoO<sub>3</sub>. Two stable  $\gamma$  and  $\gamma'$  forms and one metastable  $\gamma''$  form were observed. The relative thermal stability of the  $\gamma$  form compared with the  $\gamma'$  form has been examined by isothermal heating of a mixture of the two forms under hydrothermal conditions. Thus the low-temperature stable  $\gamma$  form transformed reversibly to the  $\gamma''$  form at  $604 \pm 3^\circ\text{C}$ , and on subsequent heating, the  $\gamma''$  form transformed irreversibly to the high-temperature stable  $\gamma'$  form in the range  $640$  to  $670^\circ\text{C}$ , depending on heating rates; however, an isothermal treatment at a temperature above  $604 \pm 3^\circ\text{C}$  brought the gradual transition of the  $\gamma''$  form into the  $\gamma'$  form.

*The Isotropic Temperature Factors of Sr(Co<sub>1-x</sub>Mn<sub>x</sub>)O<sub>3</sub> ( $x = 0, 0.1, 0.5, 0.8$  and  $1.0$ ).* H TAGUCHI, M. SHIMADA, M. KOIZUMI, AND F. KANAMARU, The Institute of Scientific and Industrial Research, Osaka University, Osaka 565, Japan. The cubic perovskite Sr(Co<sub>1-x</sub>Mn<sub>x</sub>)O<sub>3</sub> has a maximum value of  $a$ -axis at  $x = 0.3$  and a change of spin state of Co<sup>4+</sup> ion from low to high. To elucidate these properties, the isotropic temperature factor ( $B$ ) of strontium, cobalt, manganese, and oxygen atoms for  $x = 0, 0.1, 0.5, 0.8,$  and  $1.0$  have been derived from powder X-ray diffraction measurements. The isotropic temperature factor of oxygen for  $x = 0, 0.1,$  and  $1.0$  is small and that for  $x = 0.5$  and  $0.8$  is large. This fact suggests that the oxygen ion deviates from the centre of the Co-O-Mn bond in the solid solutions with  $x \geq 0.3$ . Larger CoO<sub>6</sub> octahedra and smaller MnO<sub>6</sub> octahedra, which are connected by corner sharing of oxygens of the octahedron, are distributed statistically.

*Self-Diffusion of Yttrium in Monocrystalline Yttrium Oxide: Y<sub>2</sub>O<sub>3</sub>.* R. J. GABORIAUD, Laboratoire du Métallurgie Physique, 40, Avenue du Recteur Pineau, 86022 Poitiers, France. Yttrium self-diffusion in monocrystalline yttrium oxide (Y<sub>2</sub>O<sub>3</sub>) is studied by means of the classical radiotracer technique. The few reliable diffusion data obtained in the temperature range  $1600$ – $1700^\circ\text{C}$  lead to the diffusion coefficient  $D = 3.5 \times 10^9 \exp(-72/RT)(\text{kcal/mole}) \text{m}^2 \text{sec}^{-1}$ . Experimental errors on the above numerical values are large and give, for the preexponential and energy terms, respectively,  $2 \times 10^{-7} < D_0 < 3 \times 10^{-10} \text{m}^2 \text{sec}^{-1}$  and  $62 < Q < 82 \text{kcal/mole}$ . Nevertheless these results seem in good agreement with those deduced from high-temperature and low-stress creep experiments. The theoretical aspect of self-diffusion of yttrium in Y<sub>2</sub>O<sub>3</sub> is studied in terms of point defects and lattice disorder due to the equilibrium between the oxide and its environment. This last part is confined to the restricted range of high oxygen partial pressure in which oxygen interstitials are supposed to be majority defects. Intrinsic and extrinsic diffusion behaviors are both considered on the basis of a vacancy diffusion mechanism.

*Study of Pr<sub>1-x</sub>Mn<sub>1+x</sub>O<sub>3</sub> Perovskites.* EMIL POLLERT AND ZDENĚK JIRÁK, Institute of Physics, Czechoslovak Academy of Science, Na Slovance 2, 180 40 Praha 8, Czechoslovakia. The structural and magnetic properties of the Pr<sub>1-x</sub>Mn<sub>1+x</sub>O<sub>3</sub> perovskites were studied. The increase of  $x$  (i.e., Pr/Mn < 1)